



Figure S1. IR spectra of the raw materials—Chitosan (CHS), curcumin (CUR), sodium hexametaphosphate (SHMP) and sodium tripolyphosphate (TPP). IR analysis of the raw materials confirmed their characteristic functional groups. Chitosan exhibited a broad band at $\sim 3400\text{--}3200\text{ cm}^{-1}$ corresponding to overlapping O–H and N–H stretching vibrations, along with bands at $\sim 1650\text{ cm}^{-1}$ and $\sim 1580\text{ cm}^{-1}$ assigned to amide I (C = O stretching of residual acetyl groups) and NH₂ bending (amide II), respectively. Strong absorptions in the polysaccharide fingerprint region were observed at $\sim 1150\text{ cm}^{-1}$ and $\sim 1050\text{ cm}^{-1}$, attributed to asymmetric C–O–C and C–O stretching vibrations. Sodium tripolyphosphate and sodium hexametaphosphate showed no absorption in the high-wavenumber region and were dominated by phosphate vibrations between ~ 1250 and 900 cm^{-1} , including bands at $\sim 1100\text{ cm}^{-1}$ (PO₂⁻ asymmetric stretching) and $\sim 1000\text{--}950\text{ cm}^{-1}$ (P–O–P stretching), with SHMP displaying broader features than TPP. Curcumin displayed characteristic aromatic and conjugated vibrations at ~ 1600 and $\sim 1500\text{ cm}^{-1}$, together with a band at $\sim 1275\text{ cm}^{-1}$ corresponding to enolic C–O stretching, confirming its molecular structure.